# One pot Suzuki coupling – Wittig olefination reactions Thies Thiemanna\*, Masataka Watanabeb, Yasuko Tanakaa and Shuntaro Matakaa

<sup>a</sup>Institute of Materials Chemistry and Engineering, Kyushu University, 6-1, Kasuga-koh-en, Kasuga-shi, Fukuoka 816-8580, Japan bInterdisciplinary Graduate School of Engineering, Kyushu University, 6-1, Kasuga-koh-en, Kasuga-shi, Fukuoka 816-8580, Japan

Bromoarylcarbaldehydes and bromothienylcarbaldehydes undergo Suzuki cross coupling and Wittig reactions in a one pot procedure to give olefins with extended pi systems. Dihaloarenes can be transformed in a double Wittig – / double Suzuki cross coupling reaction to yield functionalised terphenyls.

Keywords: Suzuki cross-coupling, Wittig olefination, conjugated pi-systems, terphenyls

Numerous conjugated phosphoranes are stable towards water and oxygen. Many of these stabilised phosphoranes react preferentially or exclusively with carbaldehydes, while being otherwise unreactive towards other species. This inertness to other substrates and their tolerance of reaction conditions, such as of acidity/basicity of the medium and reaction temperature, make these phosphoranes useful reagents for olefination reactions with carbaldehydes in combination with other transformations in a one-pot procedure. The authors have already shown that the Wittig olefination can be combined with a [4 + 2]-cycloaddition<sup>2</sup> of the olefinic product and have demonstrated the possibility of preparing a library of functionalised phosphoranes by transforming haloaroyl- and halohetaroylmethylidenetriphenylphosphoranes in Pd(0) catalysed C-C bond reactions, such as Suzuki reactions, Sonogashira- and Heck coupling reactions.<sup>3</sup> In this paper, the potential of a combination of Suzuki cross coupling reaction and Wittig olefination with bromoarylcarbaldehydes and bromothienylcarbaldehydes is investigated.

The reaction of either bromoarenealdehydes 2b, 2c, 2e or of bromothienylaldehydes 2a, 2f with arylboronic acids and conjugated phosphoranes gives access to compounds 4 in good yield (Scheme 1, Table 1). A two phase system was used in these reactions with DME as the organic solvent and aq. Na<sub>2</sub>CO<sub>3</sub> as the aqueous medium.

In the experiments, bistriphenylphosphinopalladium (II) dichloride [(PPh<sub>3</sub>)<sub>2</sub>PdCl<sub>2</sub>]<sup>5</sup> was utilised as pre-catalyst for the Suzuki cross-coupling reactions, where additional

triphenylphosphine (PPh3) was added as ligand. The transformation also proceeded without the addition of PPh3, but especially in the case of the two-fold Suzuki coupling reactions, better yields were achieved in the presence of PPh<sub>3</sub>.

Phosphoranes 3b, 3d and 3e are more reactive towards aldehydes than the benzoylmethylidenetriphenylphosphorane (3a) and acetylmethylidenetriphenylphosphorane (3c). Thus, 3b, 3d and 3e react with aldehydes exothermally. The evolution of heat is observed especially in their reaction in solventless systems and at high concentrations. While no evolution of heat was observed in the corresponding reaction with 3a and 3c. Accordingly, the one-pot Suzuki crosscoupling - Wittig olefination reactions necessitated longer reaction times with 3a and 3c.

Only in one case did the one-pot procedure fail, namely in the reaction of 5-bromothienylcarbaldehyde (2f) with acetylmethylidenetriphenylphosphorane (3c) and benzofuranboronic acid, where the Suzuki coupling reaction did not proceed satisfactorily and a mixture of mostly 4-(5'-bromothien-2'-yl) but-3-en-2-one and the reduced 4-(thien-2'-yl)but-3-en-2-one was produced.

Bishaloarenes and bisheteroarenes can also be subjected to a two-fold Suzuki coupling - Wittig olefination process, when 4-formylphenylboronic acid is used as a building block.<sup>6</sup> Here, terphenyls conjugated to further pi functional groups can be synthesised readily in one step. The thienyl 6c is of interest as a new class of fluorescent materials (Scheme 2,

### Scheme 1

$$Br \longrightarrow I + (HO)_{2}B \longrightarrow CHO + Ph_{3}P \longrightarrow Y \xrightarrow{(Ph_{3}P)_{2}PdCl_{2} \atop (3.5 \cdot 4.5 \text{ mol}\%)} PPh_{3}(2 \text{ equiv.})$$

$$5a \qquad 1f \qquad 3a/c \qquad 6$$

$$Y = Me, Ph$$

Scheme 2

<sup>\*</sup> Correspondence.

B(OH) <sub>2</sub> + Br	HCOMe  Weo  Yield \$4%; only E-isoner  MeO  MeO  HCOPh	1a 2f 3a Yield 72%; only E-isomer isolated (15h)  MeO	Horoman Annual San Ann
MeO	+ Ph <sub>3</sub> P=CHCO <sub>2</sub> Et 4b 3b Yreld 94%; E/Z 83/15 (10.5h) 4 Ph <sub>3</sub> P=CHCOCH <sub>3</sub> MeO 4c	MeO + CHO Ph <sub>3</sub> P O MeO + A Yield 61% (13h) WeO + A A Yield 59%; based on the purified E-isomer (9h)	MeO  + Br  - CHO + Ph <sub>3</sub> P=CHCO <sub>2</sub> Et  + Br  - CHO + Ph <sub>3</sub> P=CHCO <sub>2</sub> Et  - Weo  - 4e  - Yield 77%; ElZ 95/5 (12h)  - Yield 83%; ElZ 93/7 (12h)  - 1c  - 2c  - 3b  - 4f  - Yield 83%; ElZ 93/7 (12h)  - 1c

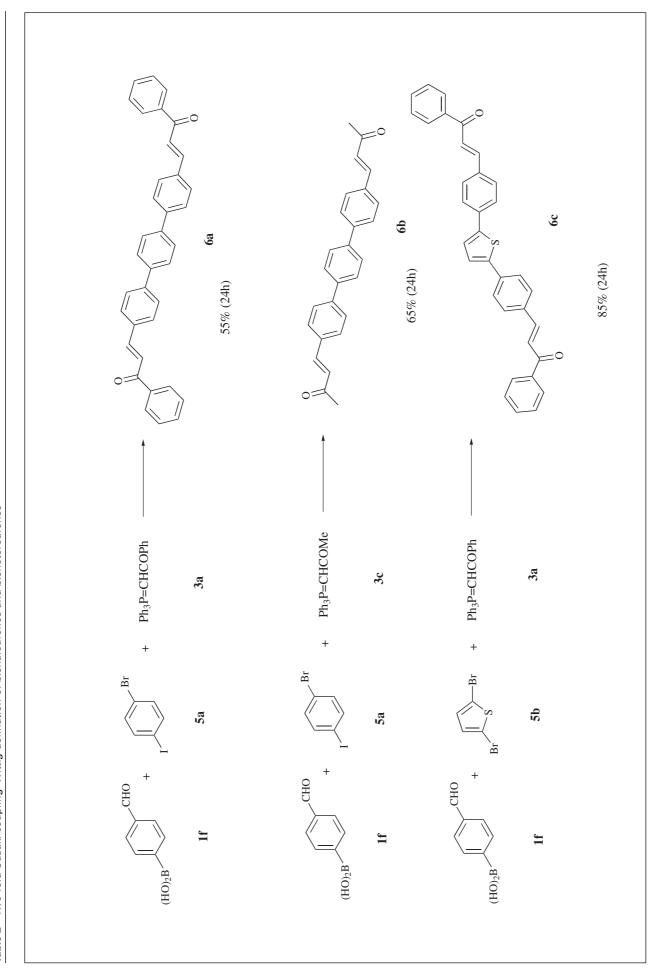


Table 2). In general, the reaction time (24h) for the two-fold reaction is longer than for the reactions described above.

Most of the products needed to be separated by column chromatography. The terphenyls **4e**, **6a** and **6b** and the quaterphenyl **4f**, however, are sparingly soluble and crystallised from the reaction mixtures. In these cases, after termination of the reaction, water was added and both the organic and the aqueous phase were filtered. The filter cake holds the products and was extracted with hot solvent (chloroform/benzene) to yield the pure product after evaporation of the solvent and crystallisation of the residue from ether. Here, a continuous extraction with a Soxhlet apparatus is of advantage.

As part of our efforts to reduce solvents and waste materials through the development of multi-step one-pot reactions in combination with the use of solventless systems, <sup>7</sup> studies are currently underway to carry out the transformations presented here in solventless systems.

### **Experimental**

Melting points were measured on a Yanaco microscopic hotstage and are uncorrected. Infrared spectra were measured with a JASCO IR-700 instrument. <sup>1</sup>H (270 MHz and 395.7 MHz) were recorded with a JEOL EX-270 and a JEOL Lambda 400 FT-NMR spectrometer, and, <sup>13</sup>C (99.45 MHz) with a JEOL Lambda 400 FT-NMR spectrometer. The chemical shifts are relative to TMS (solvent CDCl3, unless otherwise noted). Mass spectra were measured with a JMS-01-SG-2spectrometer (EI, 70 eV). UV-VIS spectra of 6c were carried out with a JASCO UV/VIS/NIR V-570 spectrophotometer, the fluorescence spectra with a Hitachi F-4500 Fluorescence Spectrophotometer. Column chromatography was carried out on Wakogel 300. All experiments were purged with argon at the start. 4-Bromothienyl-2-carbaldehyde (2d) (Aldrich), 4-bromobenzaldehyde (2b) (TCI), bistriphenylphosphinopalladium(II) dichloride [(PPh<sub>3</sub>)<sub>2</sub>PdCl<sub>2</sub>] (Aldrich) and the boronic acids (Aldrich, TCI) are commercial materials. Bromophenylbenzaldehydes 2c and 2e were synthesised by Suzuki cross coupling reaction of 4-formylphenylboronic acid (1f) with 1,2-bromoiodobenzene (Aldrich) and 1,4-bromoiodobenzene (5a), respectively. The phosphoranes 3a,8a 3b,8b 3c8a, 3d8c and 5bromothienylcarbaldehyde (2f)<sup>9</sup> were prepared according to literature procedures. Compound **4b** (Table 1) is a known compound. <sup>10</sup>

 $4\hbox{-}(4'\hbox{-}Methoxyphenyl)\hbox{-}2\hbox{-}(2''\hbox{-}benzoylethenyl) thiophene (\textbf{4a}) General$ Procedure A: A mixture of 2a (350 mg, 1.8 mmol), 1a (347 mg, 2.3 mmol), 3a (1.05 g, 2.7 mmol), bistriphenylphosphinopalladium (II) dichloride [(PPh<sub>3</sub>)<sub>2</sub>PdCl<sub>2</sub>] (45 mg, 6.4·10<sup>-2</sup> mmol), triphenylphosphine (34 mg, 0.12 mmol) in DME (8 ml) and 2M aq. Na<sub>2</sub>CO<sub>3</sub> (4 ml) was kept at 70°C for 12h. The mixture was cooled, diluted with water (50 ml) and extracted with chloroform (2 × 30 ml). The organic phase was dried over anhydrous MgSO<sub>4</sub> and concentrated in vacuo. The residue was chromatographed on silica gel (ether/hexane/ chloroform 1:4:1) to yield 4a (498 mg, 85%) as pale yellow crystals, m.p. 179°C (hexane/ether 1:1); (Found: M+ 320.0869. C<sub>20</sub>H<sub>16</sub>O<sub>2</sub>S requires M, 320.0871);  $v_{\rm max}$  (KBr)/cm<sup>-1</sup> 3090, 1658, 1596, 1509, 1280, 1256, 1199, 1177, 1012, 825, 775;  $\delta_{\rm H}$  (270 MHz, CDCl<sub>3</sub>) 3.85 (3H, s, OCH<sub>3</sub>), 6.96 (2H, d,  $^3J$  8.4 Hz), 7.38 (1H, d,  $^3J$  15.4 Hz), 7.43 (1H, bs), 7.47 - 7.63 (6H, m), 7.95  $(1H, d, {}^{3}J$  15.4 Hz), 8.02 (d, 2H, d, 3J) $^{3}J$  8.4 Hz);  $\delta_{\rm C}$  (99.45 MHz, CDCl<sub>3</sub>) 55.34, 114.31, 120.87, 122.20, 127.43, 127.70, 128.41, 128.63, 130.68, 132.78, 137.23, 138.12, 143.19, 159.28, 189.82; MS (70 eV) m/z (%): 320 (100) [M+], 289 (19), 243 (15), 215 (18). (Found: C, 74.51; H, 5.06. C<sub>20</sub>H<sub>16</sub>O<sub>2</sub>S requires C, 74.97; H, 5.03).

## Further selected physical and spectral data

 $4\text{-}(4''\text{-}Methoxybiphenylyl)buten-2-one}$  (**4c**): Colourless prisms, m.p.  $184^{\circ}\text{C}$ ; (Found: M+ 252.1552.  $C_{17}\text{H}_{16}\text{O}_2$  requires M, 252.1150);  $V_{\text{max}}$  (KBr)/cm<sup>-1</sup> 2956, 2836, 1657, 1599, 1498, 1358, 1293, 1254, 1192, 1036, 977, 840, 808;  $\delta_{\text{H}}$  (270 MHz, CDCl $_3$ ) 2.40 (3H, s CH $_3$ ), 3.86 (3H, s, OCH $_3$ ), 6.74 (1H, d,  $^3J$  16.7 Hz), 6.98 (2H, d,  $^3J$  8.4 Hz), 7.51 - 7.65 (7H, m); MS (70 eV) m/z (%) 252 (72).

*N-Phenyl-3-(4"-methoxybiphenylylmethylidene)-4(H)-maleimide* (**4d**): Colourless solid, m.p. 227°C; (Found: MH<sup>+</sup> 370.1444. C<sub>24</sub>H<sub>20</sub>O<sub>3</sub>N requires M, 370.1443); ν<sub>max</sub> (KBr)/cm<sup>-1</sup> 1768, 1706, 1647, 1598, 1498, 1381, 1249, 1173, 700, 669; δ<sub>H</sub> (270 MHz, CDCl<sub>3</sub>) 3.81 (2H, d, <sup>4</sup>*J* 2.1 Hz), 3.87 (3H, s, OCH<sub>3</sub>), 7.02 (2H, d, <sup>3</sup>*J* 8.7 Hz), 7.37 – 7.70 (11H, m), 7.77 (1H, t, <sup>4</sup>*J* 2.1 Hz); δ<sub>C</sub> (99.45 MHz, CDCl<sub>3</sub>) 34.42, 55.39, 114.46, 122.39, 126.48, 127.20, 128.20, 128.58,

129.15, 130.85, 131.04, 132.34, 135.13, 143.20, 159.85, 162.01, 170.01, 173.09; MS (FAB, 3-nitrobenzyl alcohol) m/z (%) = 370 (2.6) [MH+]; (Found: C, 76.81; H, 5.18; N, 3.72.  $C_{24}H_{19}NO_3\cdot0.25H_2O$  requires: C, 77.09; H, 5.26; N, 3.74)

Ethyl 3-(4"-methoxy-p-terphenyl)acrylate (4e): Colourless crystals, mp. 330°C\* (ether) (Found: M† 358.1570.  $C_{24}H_{22}O_{3}$  requires M, 358.1569); $v_{\rm max}$  (KBr)/cm<sup>-1</sup> 1706, 1631, 1599, 1491, 1309, 1288, 1253, 1202, 1176, 1035, 813;  $\delta_{\rm H}$  (270 MHz, CDCl<sub>3</sub>) 1.31 (3H, t,  $^3J$  7.1 Hz), 3.86 (3H, s, OCH<sub>3</sub>), 4.28 (2H, q,  $^3J$  7.1 Hz), 6.48 (1H, d,  $^3J$  15.9 Hz), 7.00 (2H, d,  $^3J$  8.6 Hz), 7.49 – 7.76 (m, 11H); MS (70 eV) m/z (%): 358 (46), 277 (48), 58 (100). \*Shows three unidentified mesophases at 260 – 307°C; at 307 – 327°C and 327–330°C.

*Ethyl 3-(quaterphenyl)acrylate* (**4f**): Colorless crystals, 346°C (ether); (Found: M<sup>+</sup> 404.1773.  $C_{29}H_{24}O_2$  requires M, 404.1776);  $V_{max}$  (KBr)/cm<sup>-1</sup> 1713, 1633, 1484, 1311, 1176, 816, 764;  $δ_H$  (270 MHz, CDCl<sub>3</sub>) 1.36 (3H, t,  $^3J$  7.0 Hz), 4.29 (2H, q,  $^3J$  7.0 Hz), 6.49 (1H, d,  $^3J$  15.9 Hz), 7.31 – 7.77 (18H, m); MS (70 eV) m/z (%): 404 (19) [M<sup>+</sup>], 358 (65), 306 (100).

Ethyl 3-(4'-phenylthien-2'-yl)isobut-2(3)-enoate (4g): m.p. 89°C (hexane/ether 1:1); (Found: M<sup>+</sup> 272.0876.  $C_{16}H_{16}O_2S$  requires M, 272.0871); $v_{max}$  (KBr)/cm<sup>-1</sup> 3094, 2984, 1704, 1614, 1265, 1242, 1184, 1122, 1109, 741;  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 1.36 (3H, t,  $^3$ J 7.0 Hz), 2.24 (3H, d,  $^4$ J 1.3 Hz), 4.28 (2H, q,  $^3$ J 7.0 Hz), 7.27 – 7.63 (7H, m), 7.87 (1H, d,  $^4$ J 1.3 Hz);  $\delta_C$  (99.45 MHz, CDCl<sub>3</sub>) 14.28, 14.35, 60.98, 123.89, 125.58, 126.37, 127.54, 128.90, 130.50, 131.37, 135.17, 139.98, 142.47, 168.45; MS (70 eV) mlz (%): 272 (100) [M<sup>+</sup>], 243 (18), 227 (21), 198 (60). (Found: C, 70.50; H, 5.88.  $C_{16}H_{16}O_2S$  requires C, 70.77; H, 5.92).

4-(4"-Methoxy-m-terphenyl)but-3-en-2-one (4h): A slowly crystallising solid; (Found: M<sup>+</sup> 328.1460;  $C_{23}H_{20}O_2$  requires M, 328.1463);  $V_{max}$  (KBr)/cm<sup>-1</sup> 3026, 2928, 2836, 1667, 1607, 1516, 1476, 1250, 1178, 909, 833;  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 2.37 (3H, s, CH<sub>3</sub>), 3.78 (3H, s, OCH<sub>3</sub>), 6.68 (1H, d,  $^3J$  16.2 Hz), 6.75 (2H, d,  $^3J$  8.3 Hz), 7.06 (2H, d,  $^3J$  8.3 Hz), 7.17 (2H, d,  $^3J$  8.1 Hz), 7.30 – 7.41 (6H, m), 7.47 (1H, d,  $^3J$  16.2 Hz); MS (70 eV) m/z (%): 328 (100) [M<sup>+</sup>], 145 (39).

 $2\cdot(4'\text{-}Methoxyphenyl)\text{-}5\cdot(but\text{-}1''\text{-}en\text{-}3''\text{-}on\text{-}1''\text{-}yl)thiophene}$  (4i): Bright yellow needles, m.p. 158°C; (Found: M+ 258.0716.  $C_{15}H_{14}O_2S$  requires M, 258.0715);  $V_{\text{max}}$  (KBr)/cm¹ 1656, 1635, 1604, 1505, 1446, 1258, 1178, 1025, 969, 824, 796;  $\delta_{\text{H}}$  (270 MHz, CDCl<sub>3</sub>) 2.34 (3H, s, CH<sub>3</sub>), 3.80 (3H, s, OCH<sub>3</sub>), 6.49 (1H, d,  $^3J$  15.7 Hz), 6.92 (2H, d,  $^3J$  8.6 Hz), 7.17 (1H, d,  $^3J$  4.0 Hz), 7.25 (1H, d,  $^3J$  4.0 Hz), 7.55 (2H, d,  $^3J$  8.6 Hz), 7.59 (1H, d,  $^3J$  15.7 Hz); MS (70 eV) m/z (%): 258 (82) [M¹], 243 (100). (Found: C, 69.68; H, 5.48.  $C_{15}H_{14}O_2S$  requires C, 69.93; H, 5.46).

 $2\text{-}[2"\text{-}(4'\text{-}Methoxyphenyl)thien\text{-}5"\text{-}yl]vinylphenylketone} \qquad \textbf{(4k)}:$  Light yellow crystals; m.p. 158°C; (Found: M+ 320.0870.  $C_{20}H_{16}O_2S$  requires M, 320.0871);  $V_{max}$  (KBr)/cm<sup>-1</sup> 2954, 1650, 1588, 1574, 1256, 1221, 1033, 964, 812, 692;  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 3.86 (3H, s, OCH<sub>3</sub>), 6.94 (2H, d,  $^3J$  8.9 Hz), 7.19 (1H, d,  $^3J$  4.5 Hz), 7.29 (1H, d,  $^3J$  14.9 Hz), 7.32 (1H, d,  $^3J$  4.5 Hz), 7.47 – 7.62 (5H, m), 7.93 (1H, d,  $^3J$  14.9 Hz), 8.02 (2H, m),  $\delta_C$  (99.45 MHz) 55.40, 114.49, 119.85, 123.20, 127.37, 127.63, 128.37, 128.56, 132.64, 133.82, 137.18, 137.44, 138.63, 147.99, 160.06, 189.80; MS (70 eV) m/z (%) = 320 (100) [M+], 243 (26).

4-(4'-Methoxyphenyl)-2-(2"-acetylethenyl)thiophene (**4L**): Pale yellow crystals, mp. 141°C. (Found: M+ 258.0714.  $C_{15}H_{14}O_2S$  requires M, 258.0715);  $V_{max}$  (KBr) cm<sup>-1</sup> 3096, 1669, 1614, 1508, 1255, 1180, 1025, 973, 825, 761;  $\delta_{H}$  (270 MHz, CDCl<sub>3</sub>) 2.35 (3H, s, CH<sub>3</sub>), 3.84 (3H, s, OCH<sub>3</sub>), 6.55 (1H, d,  ${}^3J$  15.9 Hz), 6.78 (1H, s), 6.93 (2H, d,  ${}^3J$  6.9 Hz), 7.40 (1H, bs), 7.47 (2H, d,  ${}^3J$  6.9 Hz), 7.64 (1H, d,  ${}^3J$  15.9 Hz); MS (70 eV) m/z (%): 258 (100) [M+], 243 (85). (Found: C, 69.62; H, 5.46.  $C_{15}H_{14}O_2S$  requires C, 69.93; H, 5.46%).

C, 69.62; H, 5.46.  $C_{15}H_{14}O_{2S}$  requires C, 69.93; H, 5.46%).  $2-[2^{\prime\prime\prime}-(o-Tolyl)thien-5^{\prime\prime\prime}-yl]vinylphenylketone (4m): Pale yellow oil; (Found: M† 304.0921. <math>C_{20}H_{16}OS$  requires M, 304.0922);  $V_{max}$  (KBr)/cm<sup>-1</sup> 3058, 1658, 1589, 1448, 1353, 1280, 1216, 1014;  $\delta_{H}$  (270 MHz, CDCl<sub>3</sub>) 2.47 (3H, s, CH<sub>3</sub>), 7.05 – 8.03 (12H, m), 7.89 (1H, d,  $^{3}J$  17.0 Hz); MS (70 eV) m/z (%) = 304 (70) [M†], 271 (25), 214 (100), 185 (42), 157 (19), 137 (47).

2-[6'-Fluorobiphenylyl]vinylphenylketone (4n): Colourless oil; (Found: M<sup>+</sup> 302.1104. C<sub>21</sub>H<sub>15</sub>OF requires M, 302.1107);v<sub>max</sub> (KBr)/cm<sup>-1</sup> 3060, 2924, 1663, 1604, 1484, 1450, 1332, 1215, 1015, 982, 756;  $\delta_{\rm H}$  (270 MHz, CDCl<sub>3</sub>) 7.13 – 7.65 (10H, m), 7.73 (2H, d,  $^3J$  8.2 Hz), 7.86 (1H, d,  $^3J$  15.9 Hz), 8.04 (2H, m); MS (70 eV) m/z (%) = 302 (100) [M<sup>+</sup>], 225 (13), 207 (31), 190 (32).

2,5-Bis(2'-benzoylethenylphenyl)thiophene (**6c**): General procedure B: A mixture of 4-bromo-1-iodobenzene (**5a**) (242 mg, 1.0 mmol), 4-formylphenylboronic acid (**1f**) (600 mg, 4.0 mmol), benzoylmethylidenetriphenylphosphorane (**3a**) (2.28 g, 6.0 mmol), [(PPh<sub>3</sub>)<sub>2</sub>PdCl<sub>2</sub>]

(30 mg,  $4.3\cdot10^{-2}$  mmol), and triphenylphosphine (23 mg,  $8.6\cdot10^{-2}$  mmol) in DME (7 ml) and 2M aq. Na<sub>2</sub>CO<sub>3</sub> solution (3 ml) was heated at 70°C for 24h. The cooled solution was then diluted with water and extracted with ether. Both the organic and the aqueous phase were filtered. The remaining filter cake was extracted with hot chloroform. The filtrate resulting from the hot extraction was concentrated *in vacuo*. Ether was added to the concentrated solution (100 ml). The precipitate was filtered and washed with ether to give **6c** (420 mg, 85%) as metallic-yellow crystals, mp. 214°C (ether); (Found: MH+497.1574. C<sub>34</sub>H<sub>25</sub>O<sub>2</sub>S requires MH, 497.1575). V<sub>max</sub> (KBr)/cm<sup>-1</sup> 1659, 1598, 1333, 1219, 1017, 802, 772, 686;  $\delta_{\rm H}$  (270 MHz, CDCl<sub>3</sub>) 7.41 (2H, s), 7.47 – 7.65 (8H, m), 7.67 (8H, bs), 7.82 (2H, d,  $^3J$  16.3 Hz), 8.03 (4H, d,  $^3J$  7.4 Hz); MS (FAB, 3-nitrobenzyl alcohol) m/z (%) = 497 (MH+, 4.9); UV  $\lambda_{\rm max}$  = 394.5 nm (lge 4.62, c = 10<sup>-5</sup> in CH<sub>2</sub>Cl<sub>2</sub>);  $F\lambda_{\rm max}$  = 513.8 nm (excitation at  $\lambda$  = 394 nm, c = 10<sup>-6</sup> in CH<sub>2</sub>Cl<sub>2</sub>)

Further selected physical and spectral data: 4,4"-Bis (benzoyle thenyl)-p-terphenyl (**6a**) as shiny silver flakes, m.p. 328°C (ether)\*;  $V_{max}$  (KBr)/cm<sup>-1</sup> 3062, 1662, 1640, 1602, 1340, 1311, 1286, 1219, 1013, 979, 805, 772, 688;  $\delta_{\rm H}$  (270 MHz, CDCl<sub>3</sub>) 7.50 – 7.75 (16H, m), 7.73 (4H, s), 7.87 (2H, d,  $^3J$  15.9 Hz), 8.05 (4H, d,  $^3J$  8.4 Hz); MS (FAB, 3-nitrobenzyl alcohol) m/z (%) = 491 (MH+, 0.3), 415 (MH+-C<sub>6</sub>H<sub>5</sub>, 0.8). \*shows an unidentified mesophase between 312°C and 328°C.

4,4"-Bis(acetylethenyl)-p-terphenyl (**6b**): shiny silver flakes; mp 325°C (dec.)\* (ether);  $V_{max}$  (KBr)/cm<sup>-1</sup> 1662, 1259, 979, 803;  $δ_H$  (270 MHz, CDCl<sub>3</sub>) 2.41 (6H, s, 2 CH<sub>3</sub>), 6.77 (2H, d,  $^3J$  16.2 Hz), 7.56 (2H, d,  $^3J$  16.2 Hz), 7.64 (4H, d,  $^3J$  8.4 Hz), 7.70 (4H, d,  $^3J$  8.4 Hz), 7.72 (4H, s); MS (FAB, 3-nitrobenzyl alcohol) m/z (%) = 367 (MH<sup>+</sup>, 1.9). \* shows unidentified mesophase between 296°C and 325°C.

Received 2 July 2004; accepted 12 September 2004 Paper 04/2621

#### References:

- O.I. Kolodiazhnyi, Phosphorus Ylides Chemistry and Application in Organic Synthesis, Wiley-VCh, Weinheim, 1999.
- 2 For a one-pot Wittig-olefination, Diels Alder reaction, see: T. Thiemann, D. Ohira, Y.Q. Li, T. Sawada, S. Mataka, K. Rauch, M. Noltemeyer, and A. de Meijere, J. Chem. Soc., Perkin Trans. 1, 2000, 2968.
- 3 (a) T. Thiemann, K. Umeno, D. Ohira, E. Inohae, T. Sawada, and S. Mataka, New J. Chem. 1999, 23, 1067; (b) T. Thiemann, K. Umeno, E. Inohae, and S. Mataka, Rep. Inst. Adv. Mat. Study, 2000, 14, 17; Chem. Abstr., 2000, 133, 335380; c) T. Thiemann, K. Umeno, J. Wang, K. Arima, M. Watanabe, Y. Tabuchi, Y. Tanaka, H. Gorohmaru, and S. Mataka, J. Chem. Soc., Perkin Trans. 1, 2002, 2090.
- (a) N. Miyaura and A. Suzuki, Chem. Rev., 1995, 95, 2457;
   (b) A. Suzuki, J. Organomet. Chem., 1999, 576, 147.
- For the use of (Ph<sub>3</sub>P)<sub>2</sub>PdCl<sub>2</sub> in Suzuki cross-coupling reactions in the synthesis of biaryls and bishetraryls, see: (a) A. Suzuki, *Cross-Coupling of Organoboron Compounds with Organic Halides* In *Metal Catalysed Cross-Coupling Reactions*, F. Diederich, P. J. Stang, (eds), VCH-Wiley, pp. 49; (b) M. Melucci, G. Barbarella, and G. Sotgiu, *J. Org. Chem.*, 2002, 67, 8877.
- 6 For the use of formylarylboronic acids as building block, also see: T. Thiemann, M. Watanabe, Y. Tanaka, and S. Mataka, New J. Chem., submitted.
- 7 T. Thiemann, M. Watanabe, Y. Tanaka, and S. Mataka, *New J. Chem.*, 2004, **28**, 578.
- (a) F. Ramirez and S. Dershowitz, *J. Org. Chem.*, 1957, 22, 41; (b)
   Wm. J. Considine, *J. Org. Chem.*, 1962, 27, 647; (c) R. F. Hudson and D. A. Chopard, *Helv. Chim. Acta*, 1963, 46, 2178.
- J. Eras, C. Galvez and F. Garcia, J. Heterocycl. Chem., 1984, 21, 215.
- (a) R.N. MacCoss, E.J. Balskus, and S.V. Ley, *Tetrahedron Lett.*,
   2003, 44, 7779; (b) A. G.M. Barett, S.M. Cramp, R.S. Richards,
   and F.J. Zecri, *Org. Lett.*, 1999, 1, 573.